REVERSAL CHARACTERISTICS **NICKEL-HYDROGEN CELI**

C. LURIE TRW SPACE AND ELECTRONICS GROUP REDONDO BEACH, CALIFORNIA THE 1993 NASA AEROSPACE BATTERY WORKSHOP **ALABAMA SPACE AND ROCKET CENTER** NOVEMBER 16 - 18, 1993

NICKEL-HYDROGEN CELL REVERSA CHARACTERISTICS

DIRECTED TOWARDS DEVELOPMENT OF A HIGH CURRENT NICKEL-HYDROGEN CELL REVERSAL CHARACTERISTICS ARE BEING STUDIED AS PART OF A TRW PROGRAM BATTERY CELL BYPASS SWITCH.

CELL BYPASS SWITCH

OPEN CIRCUIT FAILURE MODE CONSIDERED CREDIBLE FOR NICKEL-HYDROGEN CELLS

BYPASS PROTECTION TRADITIONALLY PROVIDED BY DIODES

DIODE APPROACH IS POWER LIMITED

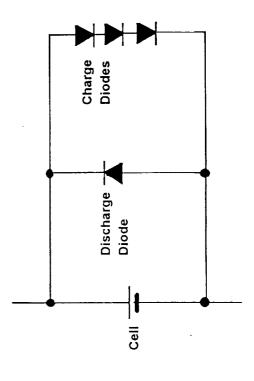
THERMAL DISSIPATION

 UNAVAILABILITY OF LARGE FLIGHT QUALIFIED DIODES

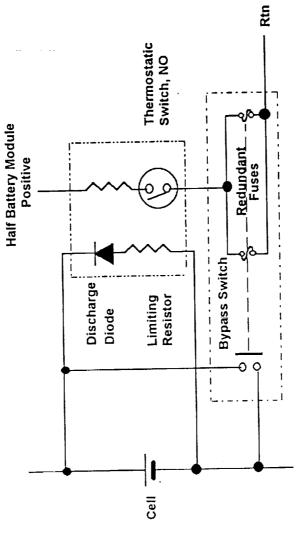
BYPASS WITH A SWITCH IS PREFERRED FOR LARGE CELLS

LOWER DISSIPATION

. LIGHTER



BYPASS SWITCH APPROACH



Passive fuse wire actuated bypass switch

Charge Diodes Redundant Fuses Discharge Diode **Bypass Switch**

Thermostatically actuated bypass switch

Nickel-Hydrogen Technologies Session

REVERSAL CHARACTERISTICS NICKEL-HYDROGEN CELI

CHARACTERIZATION OF THE SWITCH INCLUDES UNDERSTANDING

. HOW LONG IT WILL TAKE TO ACTUATE

AT WHAT VOLTAGE IT WILL ACTUATE

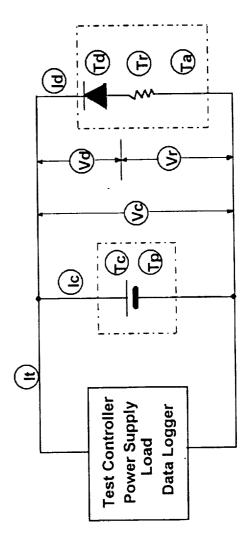
CHARACTERISTICS OF THE FUSE LINKS, DIODE, RESISTOR, AND THERMOSTAT ARE EASILY DETERMINED REVERSAL CHARACTERISTICS OF NICKEL HYDROGEN CELLS MUST ALSO BE KNOWN IF THE SWITCH ACTUATION CHARACTERISTICS ARE TO DETERMINED

EXPERIMENTAL

DETERMINE NICKEL HYDROGEN CELL REVERSAL VOLTAGE TRAJECTORIES

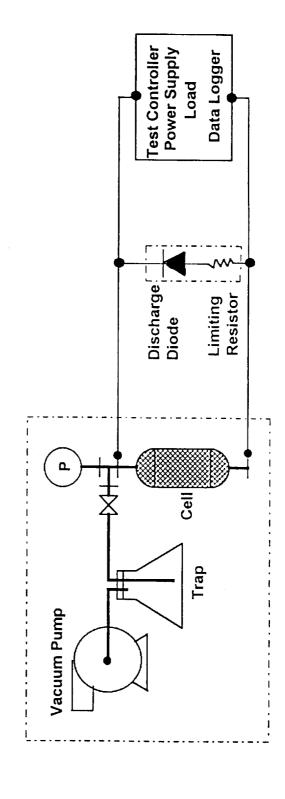
- HYDROGEN AND NICKEL PRECHARGED CELLS
- NOMINAL AND MINIMUM RATES
- CELL LEAKING INTO VACUUM (SPACE)

TEST INSTRUMENTATION



EXPERIMENTAL (CONT'D)

- C EP RNH 65-17 CELL, IN THERMAL SLEEVE, ON COLD PLATE AT 5 \pm 2 DEG
- ZIRCAR SEPARATOR, WALL WICK
- BACK-TO-BACK STACK CONFIGURATION
- LEAK TO SPACE SIMULATED BY VENTING CELL TO VACUUM



NICKEL-HYDROGEN CELL CHEMISTRY DISCHARGE/REVERSAL

DISCHARGE

Nickel
$$2 \text{ NiOOH} + 2 \text{ HOH} + 2 \text{ e} \rightarrow 2 \text{ Ni(OH)}_2 + 2 \text{ OH}^-$$

Hydrogen $H_2 + 2 \text{ OH}^- - 2 \text{ e} \rightarrow 2 \text{ HOH}$
Total $2 \text{ NiOOH} + H_2 \rightarrow \text{Ni(OH)}_2$

= +1.3v

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+0.8v

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 $E^{\circ} = +0.5v$

OVERDISCHARGE (REVERSAL) WITH HYDROGEN PRECHARGE

Nickel Hydrogen

$$2 \text{ HOH} + 2e \rightarrow \text{H}_2 + 2 \text{ OH}^-$$

 $\text{H}_2 + 2 \text{ OH}^- - 2e \rightarrow 2 \text{ HOH}$

 $E^{\circ} = -0.8v$

+0.8

H

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No net reaction; H2 recombines

OVERDISCHARGE (REVERSAL) WITH NICKEL PRECHARGE **NICKEL-HYDROGEN CELL CHEMISTRY**

ACTIVE NICKEL PRECHARGE PRESENT

 $E^{\circ} = +0.5v$

 $2 \text{ Ni(OH)}_2 + 2 \text{ OH}^-$

1

 $1/2 O_2 + HOH$

$$2 \text{ Ni(OH)}_2 + 1/2 \text{ O}_2$$

 $1/2 O_2 + HOH$

1

2 OH⁻ - 2 e

2 OH⁻ + H₂

1

2 HOH + 2 e

ACTIVE NICKEL PRECHARGE EXHAUSTED

$$E^{\circ} = -1.2v$$

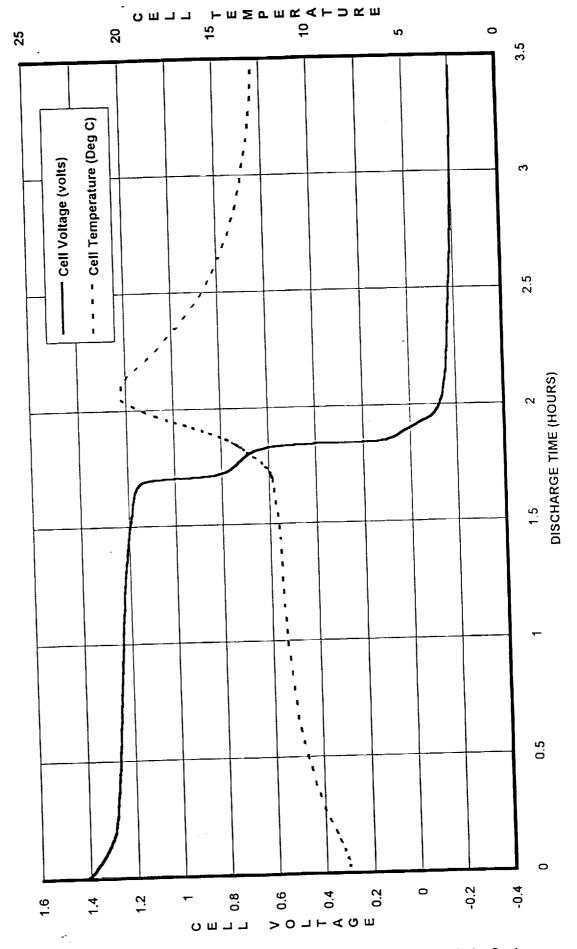
1/2 O₂ + H₂
$$\rightarrow$$

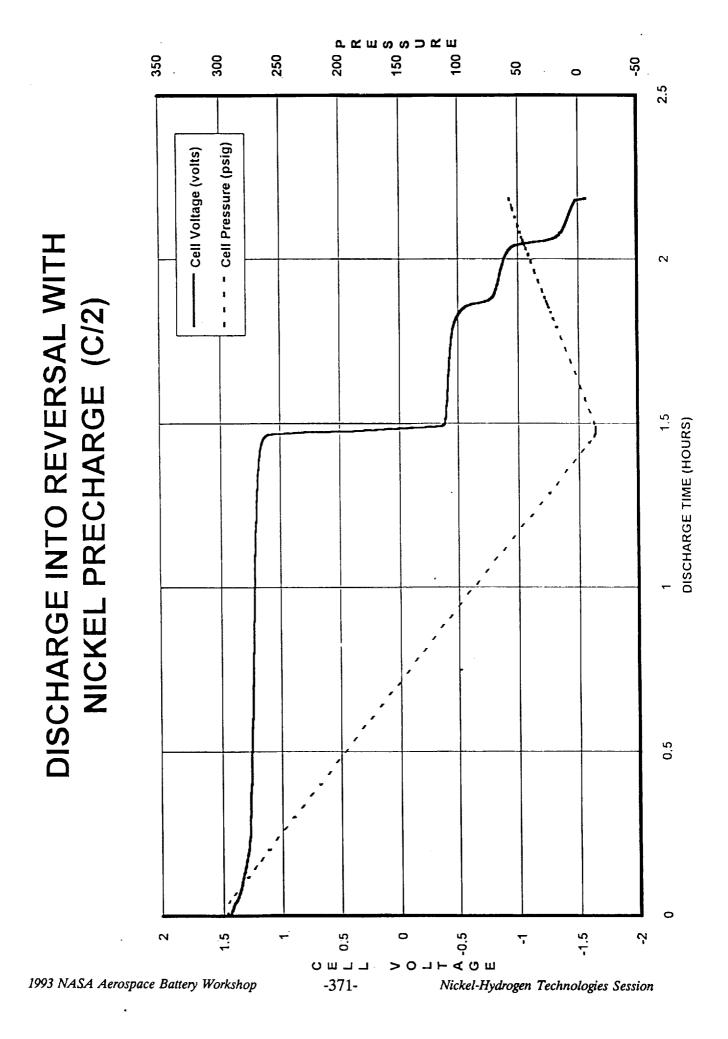
$$^{1}_{2}$$
 + $^{1}_{2}$ \rightarrow

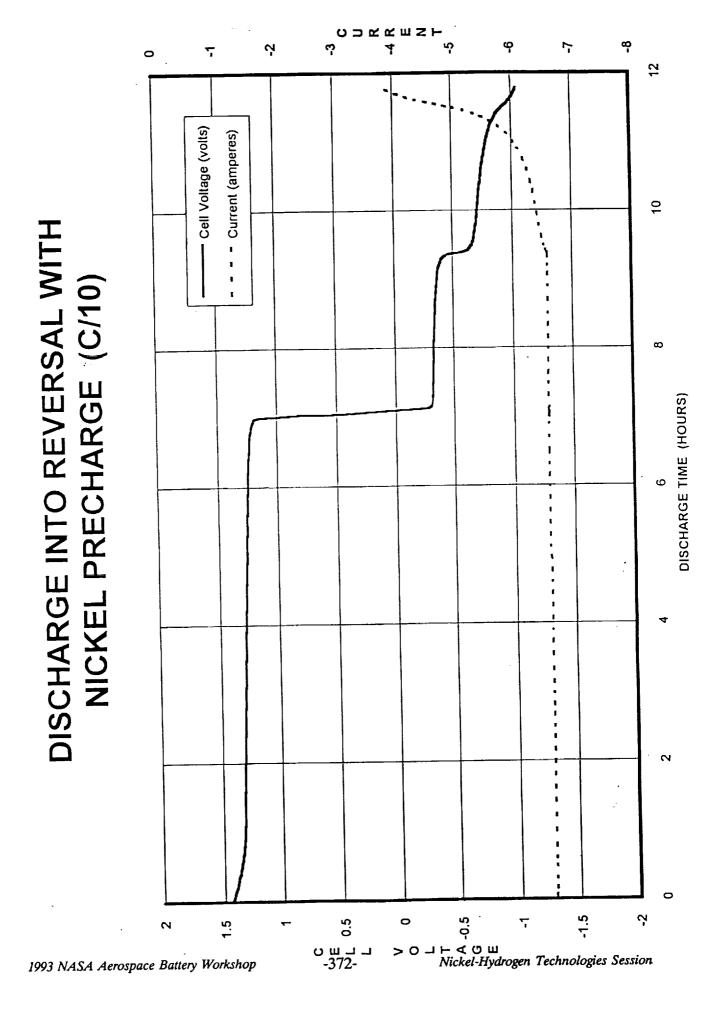
Hydrogen

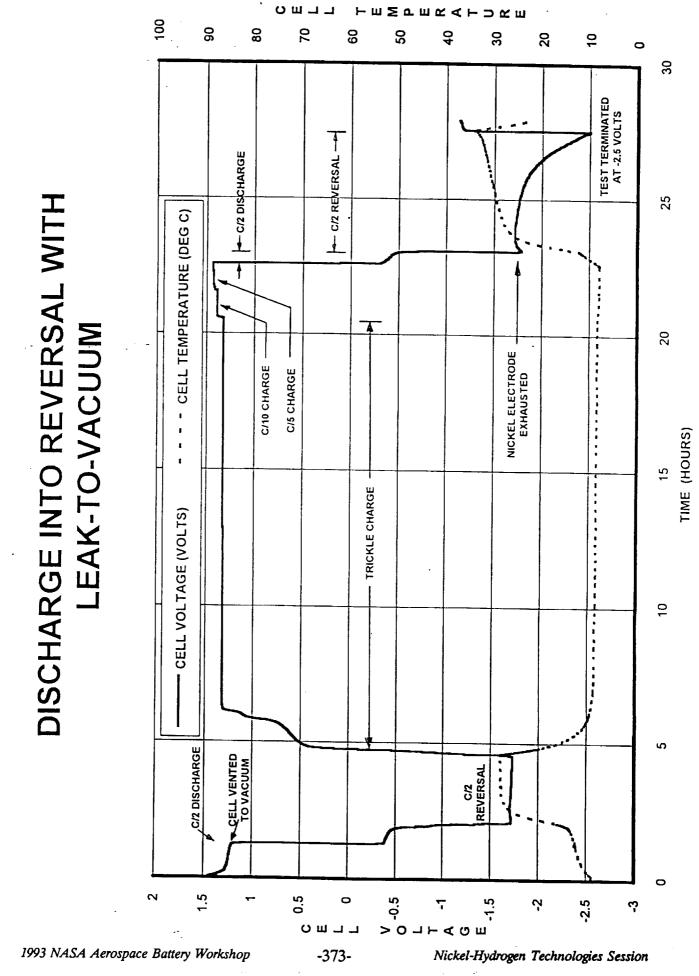
Nickel

DISCHARGE INTO REVERSAL WITH HYDROGEN PRECHARGE (C/2)



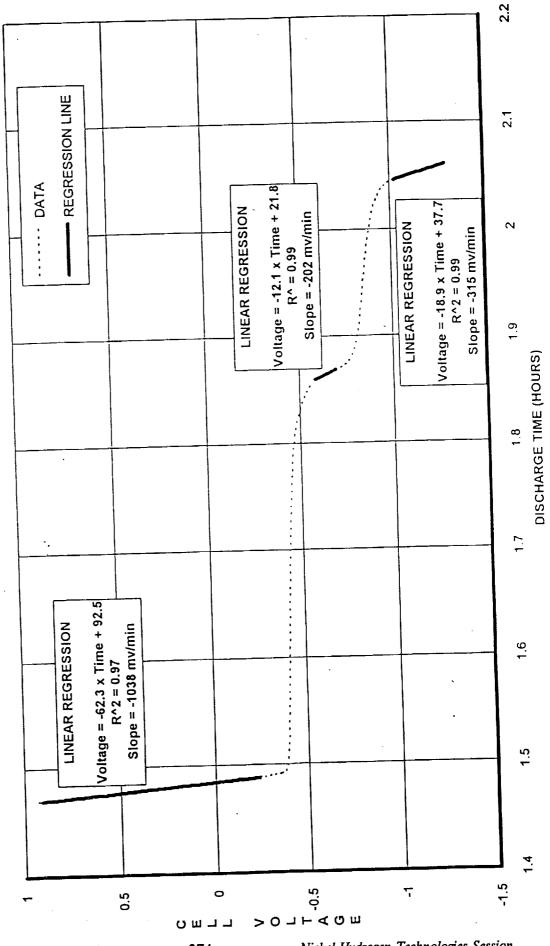


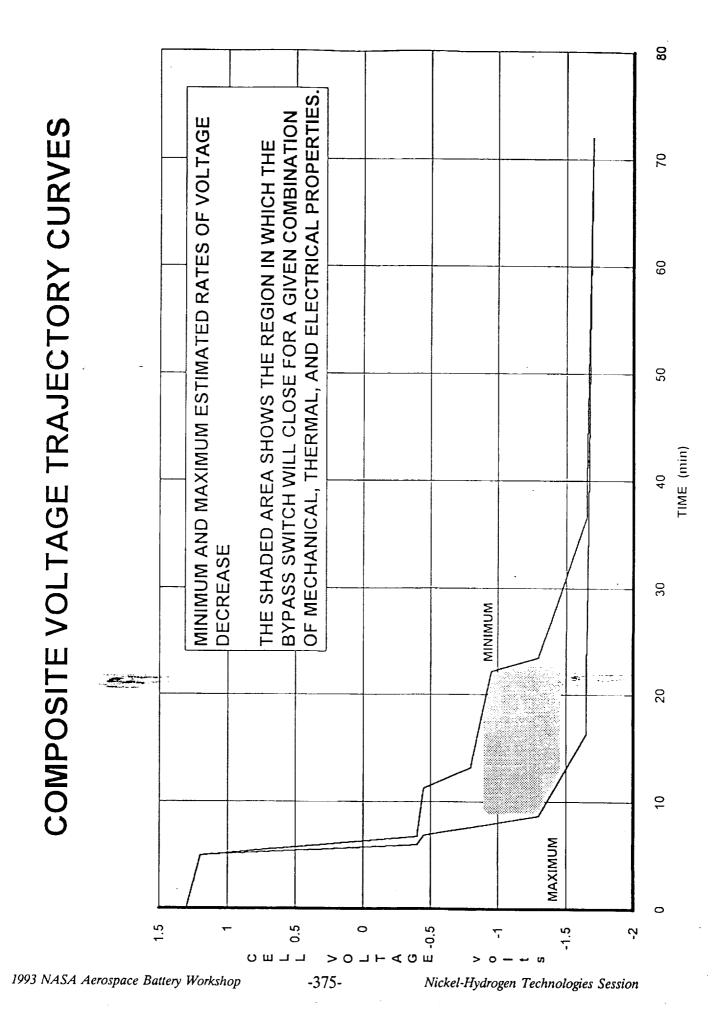




VOLTAGE TRAJECTORY END OF DISCHARGE AND REVERSAL

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